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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.	
10/773,371	02/06/2004	Keiji Ohbayashi	02126D/HG	9785	
1933	7590 11/08/2005		EXAM	EXAMINER	
•	HOLTZ, GOODMAN	PARKER, FREDERICK JOHN			
220 5TH AVE FL 16 NEW YORK, NY 10001-7708			ART UNIT	PAPER NUMBER	
,			1762		

DATE MAILED: 11/08/2005

Please find below and/or attached an Office communication concerning this application or proceeding.

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٥٠	Application No.	Applicant(s)	
Advisory Action	10/773,371	OHBAYASHI ET AL	
Before the Filing of an Appeal Brief	Examiner	Art Unit	
	Frederick J. Parker	1762	
The MAILING DATE of this communication appe	ears on the cover sheet with the c	correspondence add	ress
THE REPLY FILED 27 October 2005 FAILS TO PLACE THIS	APPLICATION IN CONDITION FO	R ALLOWANCE.	
 The reply was filed after a final rejection, but prior to or o this application, applicant must timely file one of the folioplaces the application in condition for allowance; (2) a No (3) a Request for Continued Examination (RCE) in comp following time periods: a) The period for reply expiresmonths from the mailing of the period for reply expires on: (1) the mailing date of this Advevent, however, will the statutory period for reply expire later the Examiner Note: If box 1 is checked, check either box (a) or (b) MONTHS OF THE FINAL REJECTION See MEEP 706 07/6 	owing replies: (1) an amendment, a otice of Appeal (with appeal fee) in liance with 37 CFR 1.114. The replate of the final rejection. Prisory Action, or (2) the date set forth in the an SIX MONTHS from the mailing date of the ONLY CHECK BOX (b) WHEN THE FI	ffidavit, or other evide compliance with 37 (by must be filed withing a final rejection, whichever the final rejection.	ence, which CFR 41.31; or n one of the er is later. In no
MONTHS OF THE FINAL REJECTION. See MPEP 706.07(f Extensions of time may be obtained under 37 CFR 1.136(a). The date on been filed is the date for purposes of determining the period of extension at CFR 1.17(a) is calculated from: (1) the expiration date of the shortened state above, if checked. Any reply received by the Office later than three month earned patent term adjustment. See 37 CFR 1.704(b). NOTICE OF APPEAL 2. The Notice of Appeal was filed on A brief in com	which the petition under 37 CFR 1.136(a and the corresponding amount of the fee. atutory period for reply originally set in the s after the mailing date of the final rejection	The appropriate extension final Office action; or (2) on, even if timely filed, materials	on fee under 37 as set forth in (b) ay reduce any
of filing the Notice of Appeal (37 CFR 41.37(a)), or any e Since a Notice of Appeal has been filed, any reply must be AMENDMENTS 3. The proposed amendment(s) filed after a final rejection, (a) They raise new issues that would require further co (b) They raise the issue of new matter (see NOTE below (c) They are not deemed to place the application in be	be filed within the time period set for but prior to the date of filing a brie ensideration and/or search (see NO bw);	orth in 37 CFR 41.37(f, will <u>not</u> be entered l TE below);	a). because
appeal; and/or (d) They present additional claims without canceling a NOTE: (See 37 CFR 1.116 and 41.33(a))			
 4. The amendments are not in compliance with 37 CFR 1.1 5. Applicant's reply has overcome the following rejection(s 		ompliant Amendment	(PTOL-324).
 6. Newly proposed or amended claim(s) would be a the non-allowable claim(s). 		, timely filed amendm	ent canceling
7. For purposes of appeal, the proposed amendment(s): a) how the new or amended claims would be rejected is pro The status of the claim(s) is (or will be) as follows: Claim(s) allowed: Claim(s) objected to: Claim(s) rejected: Claim(s) withdrawn from consideration:		ill be entered and an	explanation of
AFFIDAVIT OR OTHER EVIDENCE			
8. The affidavit or other evidence filed after a final action, be because applicant failed to provide a showing of good an and was not earlier presented. See 37 CFR 1.116(e).	ut before or on the date of filing a N id sufficient reasons why the affida	Notice of Appeal will <u>r</u> vit or other evidence i	ot be entered is necessary
9. The affidavit or other evidence filed after the date of filing entered because the affidavit or other evidence failed to a showing a good and sufficient reasons why it is necessar	overcome <u>all</u> rejections under appe ry and was not earlier presented. S	al and/or appellant fa See 37 CFR 41.33(d)(ils to provide a 1).
10. The affidavit or other evidence is entered. An explanation	on of the status of the claims after e	entry is below or attac	:hed.

U.S. Patent and Trademark Office PTOL-303 (Rev. 7-05)

🕅 Other: ___

REQUEST FOR RECONSIDERATION/OTHER

See Continuation Sheet.

11. The request for reconsideration has been considered but does NOT place the application in condition for allowance because:

12. Note the attached Information Disclosure Statement(s). (PTO/SB/08 or PTO-1449) Paper No(s

Continuation of 11. does NOT place the application in condition for allowance because: it fails to change the Examiner's position. To be fair to Applicants, the Examiner and a second process Examiner considered Applicants' arguments and Figure A. It was agreed that fig. A fails to show what Applicants say it does. The second Examiner agreed with this Examiner's interpretation, and pointed out apparently so does the reference which Applicants cited. One notes that the meanings of the reference set forth on page 6 of Applicants Remarks are essentially the same as those set forth by the Examiner on page 8- bridging 9 of his Final Office Action. The Examiner stands by the position that A II cannot be"a constant drying rate period"; if the solid curve is the rate of total moisture plotted against time, the slope is changing (downward) and therefore so is the rate, i.e. it cannot be "constant". Thus Applicants confusing and contradictory arguments are not persuasive, and respectfully submits that Applicants own reference supports, rather than refutes, the Examiner's interpretation of the terms and curves. The Examiner also points out new Fig. B is consistent with the Examiner's reasoning, yet did not appear to be explained in the Remarks relative to the Examiner's position, hence its meaning is unclear. However, the Examiner submits for the record evidence from the M-H Encycl. of Science and Technology, wherein figure 1 shows dark curve 1 which is the drying rate curve for a thin layer (coating) which depicts in curve form what the Examiner has consistently held: an increased (upward sloping curve) rate, followed by a long constant drying rate (flat section) and a falling rate as deposited by the dashed right, downward sloping end of the curve.

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Science & Technology

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New York San Francisco Washington, D.C. Auckland Bogotá Caracas Lisbon London Madrid Mexico City Milan Montreal New Delhi San Juan Singapore Sydney Tokyo Toronto Design and S Marine Staff,

usually water, m equipment emove liquids al dewatering lecantation or ch no change experienced. dration, which ith the drying plies removal mical change. n the chemical nemicals of all raterials, foods, lustrial wastes, iquids at rates er hour to 10 g temperatures C), or as low Dryers range ay dryers with nd 30 ft (9 m) nay be in the slurries, pastes, pers, or sheets. onvective heat ed surfaces, by ig. In general, ds (that is, the of gases are and adsorption are performed ned distillation ; (for gases and

of solids, the rm. Thus, even m, the problem is classed under its of dry solids many instances,

v compression.

g of solids is a drying is done eral case, a wet ter were present ace evaporation dled free water n an open panaring this initial erred to as the ration occurs at ant of the solid solved salts will

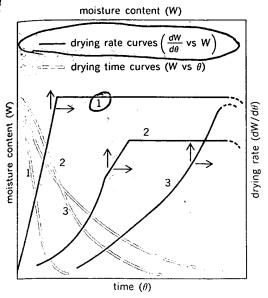


Fig. 1. Drying-time and drying-rate curves illustrating the general problem of drying (1) Curves typical of a layer of thin material with most of the drying in the constant rate. (2) A more general case in which two stages in the falling-rate period occur. Typical of granular materials. (3) A case in which no constant rate occurs. Typical of homogeneous and colloidal materials such as soap, gelatin, and viscous solutions.

cause the evaporation rate to be less than that of pure water. Nevertheless, this lower rate can still be constant during the first stages of drying.

A fundamental theory of drying depends on a knowledge of the forces governing the flow of liquids inside solids. Attempts have been made to develop a general theory of drying on the basis that liquids move inside solids by a diffusional process. However, this is not true in all cases. In fact, only in a limited number of types of solids does true diffusion of liquids occur. In most cases, the internal flow mechanism results from a combination of forces which may include capillarity, internal pressure gradients caused by shrinkage, a vapor-liquid flow sequence caused by temperature gradients, diffusion, and osmosis. Because of the complexities of the internal flow mechanism, it has not been possible to evolve a generalized theory of drying applicable to all materials. Only in the drying of certain bulk objects such as wood, ceramics, and soap has a significant understanding of the internal mechanism been gained which permits control of product quality.

Most investigations of drying have been made from the so-called external viewpoint, wherein the effects of the external drying medium such as air velocity, humidity, temperature, and wet material shape and subdivision are studied with respect to their influence on the drying rate. The results of such investigations are usually presented as drying rate curves, and the natures of these curves are used to interpret the drying mechanism. **Figure 1** shows a series of typical drying-rate curves.

The constant-rate period of drying when heat is

supplied by convection is susceptible to theoretical and analytical treatment because it is essentially independent of the solid material. When drying is accomplished by heat transfer from hot gases, which also remove the evolved vapors, the constant rate may be expressed in terms of heat-transfer rates or mass-transfer rates.

A constant rate of evaporation at the surface of the solid maintains the surface at a constant temperature, which, in the absence of other heat effects, is very nearly the wet-bulb temperature of the air. This temperature may range from 70 to 130°F (21 to 54°C) for convection drying, depending on the temperature and humidity of the air and on radiation. This so-called wet-bulb cooling effect is one reason why heat-sensitive solids can be dried in air at temperatures well above the decomposition temperature of the solid.

The magnitude of the constant rate can vary widely, depending on the degree of subdivision of the material, that is, the manner in which the material is exposed to the drying air. Thus, the rate of drying in spray dryers can be several hundred-thousand-fold greater than the rates in tray dryers.

A number of empirical expressions based on experimental studies have been developed for estimating the constant rate for different physical configurations of the wet material.

When materials are dried in contact with hot surfaces, termed indirect drying, the air humidity and air velocity may no longer be significant factors controlling the rate. The "goodness" of the contact between the wet material and the heated surfaces,

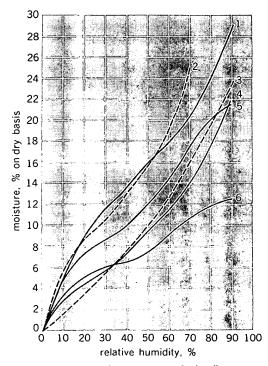


Fig. 2. Equilibrium moisture content of miscellaneous organic materials at 70°F (21°C): 1, leather; 2, tobacco; 3, soap; 4, wood; 5 catgut; 6, glue.